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Unit 12: Electrochemistry-Key Regents Chemistry '14-'15 Mr. Murdoch Page 13 of 61 Key e. Solving for X, we get +6, so the oxidation number for sulfur in the polyatomic ion (SO 4-2) is +6, and so sulfur will be +6 in ZnSO 4. *Check the Periodic Table to see if sulfur has a +6 charge listed; it does!

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