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(order doesn't matter) P_k (= binomial probability of k successes in n trials p = probability of a success $-p$ = probability of failure k = number of successes n = number of trials. Independent Events. $P(A \text{ and } B \text{ and } C) = P(A)$.

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(Exam Equation Sheet) Control
Mass (no mass flow across
system boundaries) Conservation

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of mass: $\dot{m} = \rho \dot{V}$.

Conservation of energy (1st Law):

$$\dot{m}(h_1 - h_2) = \dot{Q} - \dot{W} = \dot{Q} + \dot{W}_{\text{shaft}} + \dot{W}_{\text{flow}} = \dot{Q} + \dot{W}_{\text{shaft}} + \dot{m}(h_2 - h_1)$$

Basic Thermodynamic Formulas
(Exam Equation Sheet)

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Type or Name General Formula

1. Alkane or paraffin (also saturated hydrocarbons)
2. Alkene or olefin (unsaturated hydrocarbons)
- 3.

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solution, ($[]$ refers to molarity)
where reaction is in the gas

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NCEES FE exam information

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This article will help you improve your Excel spreadsheet skills and will provide you with tools to solve common chemical engineering problems. Ever since Lotus 1-2-3 and the IBM PC became popular in the early 1980s, chemical engineers have

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This list gives you some of the most common conversion factors you need in thermodynamics.

Acceleration: $1 \text{ m/s}^2 = 100 \text{ cm/s}^2$

Area: $1 \text{ m}^2 = 10^4 \text{ cm}^2 = 10^6 \text{ mm}^2$

Density: $1 \text{ g/cm}^3 = 1$

$\text{kg/L} = 1,000 \text{ kg/m}^3$

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heat, work, internal energy,
enthalpy: $1 \text{ kJ} = 1,000 \text{ J} = 1,000$
 $\text{N}\cdot\text{m} = 1 \text{ kPa}\cdot\text{m}^3$. $1 \text{ kJ/kg} = 1,000$
 m^2/s^2 .

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