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calcium trioxonitrate (V); 2.

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This question was attempted by majority of the candidates and the performance was fair. In (a)(i)(ii) and (iii) most candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

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When a given mass of CaCO3 was heated, 0.25 dm3 of a gas is collected at 250 C and at a pressure of 120 kNm-2. (i) Write an equation for the reaction. (ii) Calculate the mass of the:

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